IN THE SPECIFICATION:

Please amend the paragraph beginning at page 17, line 8 as follows:

Particularly preferable examples of [[R^6]] $\underline{R^{6a}}$ and [[R^{11}]] $\underline{R^{11a}}$ include groups represented by the formula -CH_2R^{12} wherein R^{12} is a straight- or branched-chain C_{1-9} alkyl group in which at least one hydrogen atom is substituted by fluorine. Preferable examples of R^{12} include straight- or branched-chain C_{1-6} alkyl groups in which at least one hydrogen atom is substituted by fluorine, such as fluoromethyl, difluoromethyl, trifluoromethyl, perfluoroethyl, perfluoropropyl, $CF_3CF_2(CH_2)_5$, HCF_2CF_2, $H(CF_2)_4$, $H(CF_2)_6$, $(CF_3)_2CH$, CF_3CHFCF_2, etc.

Please amend the paragraph beginning at page 18, line 9 and bridging to page 19, line 6 as follows:

X_1^-, X_2^-, X_3^- and X_4^- each represent an anionic moiety of the individual starting organic salts. The anionic moiety is a conjugate base of a Brönsted acid. Examples of such Brönsted acids include Brönsted acids with strong acidity, such as sulfuric acid; monomethyl sulfate, monoethyl sulfate and like sulfuric acid monoesters; methansulfonic acid, ethanesulfonic acid, chlorosulfonic acid, fluorosulfonic acid, benzenesulfonic acid, toluenesulfonic acid, nitrobenzenesulfonic acid, trichloromethanesulfonic acid, acids represented by the formula $Rf'SO_3H$ wherein Rf' is a polyfluoroalkyl group, and like sulfonic acids; sulfonimides represented by the formula $(RfSO_2)_2NH$ or $(RfSO_2)(Rf'SO_2)NH$ wherein Rf and Rf' are different and each represents a polyfluoroalkyl group; formic acid, acetic acid, butyric acid, valeric acid, trifluoroacetic acid, perfluorobutyric acid,

perfluorooctanoic acid, 3H-octafluorobutyric acid, trichloroacetic acid and like carboxylic acids;

$HB(OCOCF_3)_4$, $HB(OCOC_2F_5)_4$, $HBPh_4$, $HB(C_6F_5)_4$, $HB(p\text{-}CF_3C_6H_4)_4$, ~~$HB[3,5\text{-}(CF_3)_2C_6H_3]$~~

$HB[3,5\text{-}(CF_3)_2C_6H_3]_4$, $HC(SO_2CF_3)_3$, $HC(SO_2C_2F_5)_3$ and like organic acids; HBF_4, HPF_6, $HSbF_6$,

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$HAsF_6$, $HBCl_4$, $HBCl_3F$, $HSbCl_6$, $HSbCl_5F$, $HClO_4$, HNO_3, $HAlCl_4$, HAl_2Cl_7 and like inorganic acids;

etc.